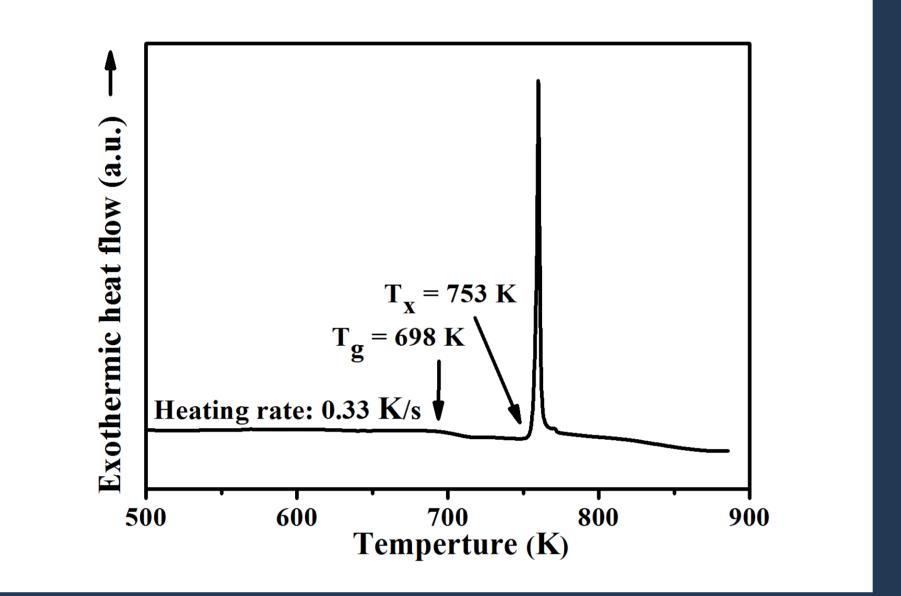
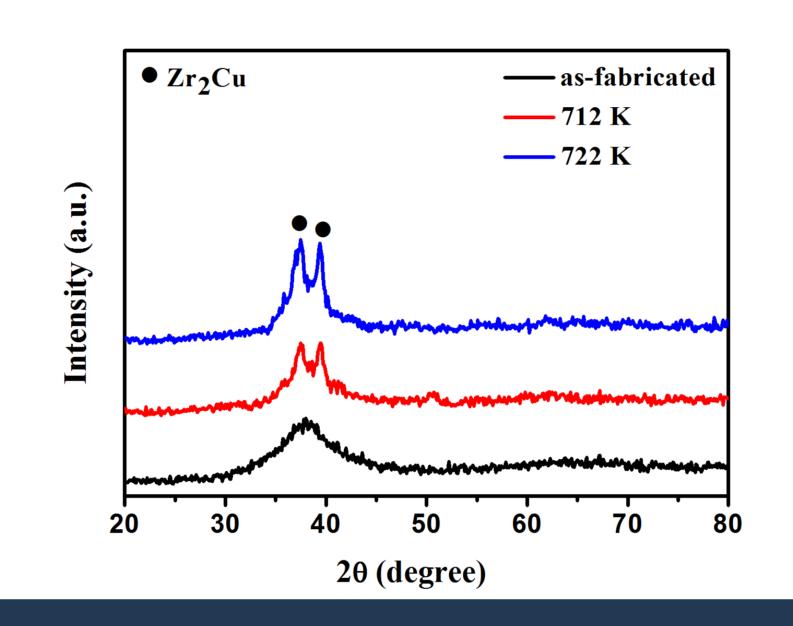
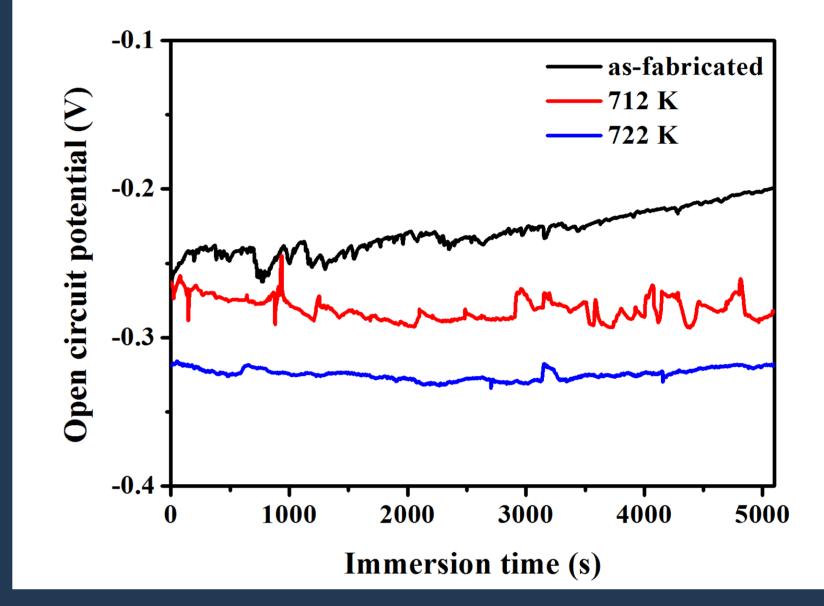


modulus, bulk metallic glasses (BMGs) have attracted interests and make them potential candidates for bioimplants. Recently, some studies pointed out that the different corrosion behavior for metallic glasses with different degrees of partial crystallization. In this study, we examine this effect in the  $Zr_{53}Cu_{30}Ni_9Al_8$  amorphous melt spun ribbons in simulated body fluid (SBF). This work would give better systematic understanding of electrochemical responses of the amorphous and nanocrystalline  $Zr_{53}Cu_{30}Ni_9Al_8$  alloys with different degrees of crystallization in SBF and would establish the clear profile of their corrosion behavior.



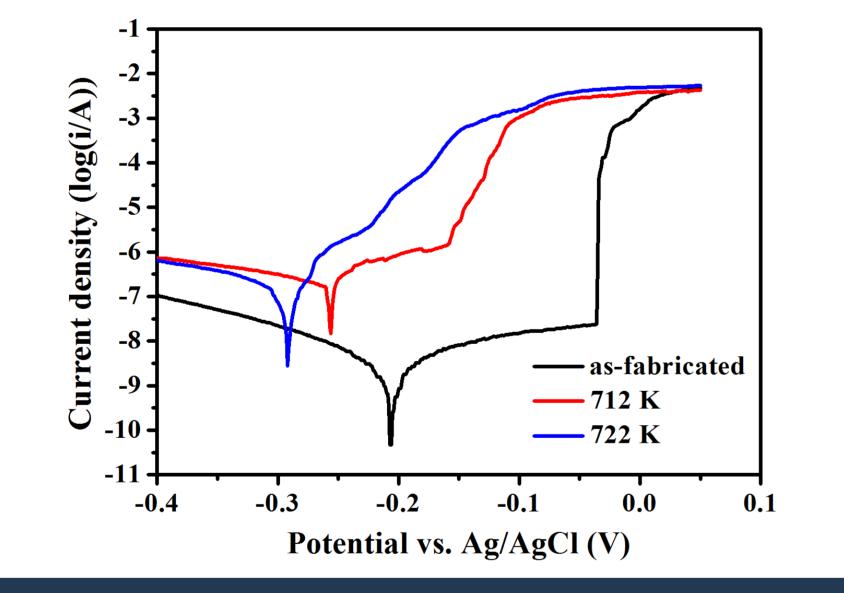






**Figure 1** DSC curve of as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  **Figure 2** XRD patterns of as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  metallic glasses.

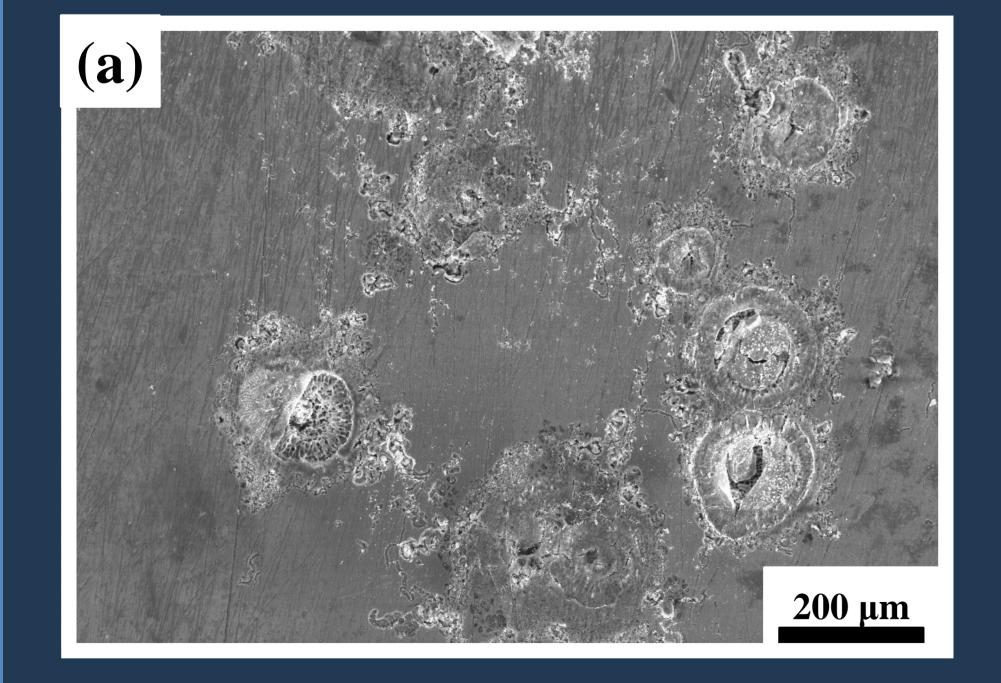
**Figure 3** Open circuit potential of as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  metallic glasses and its partial crystalline alloy.

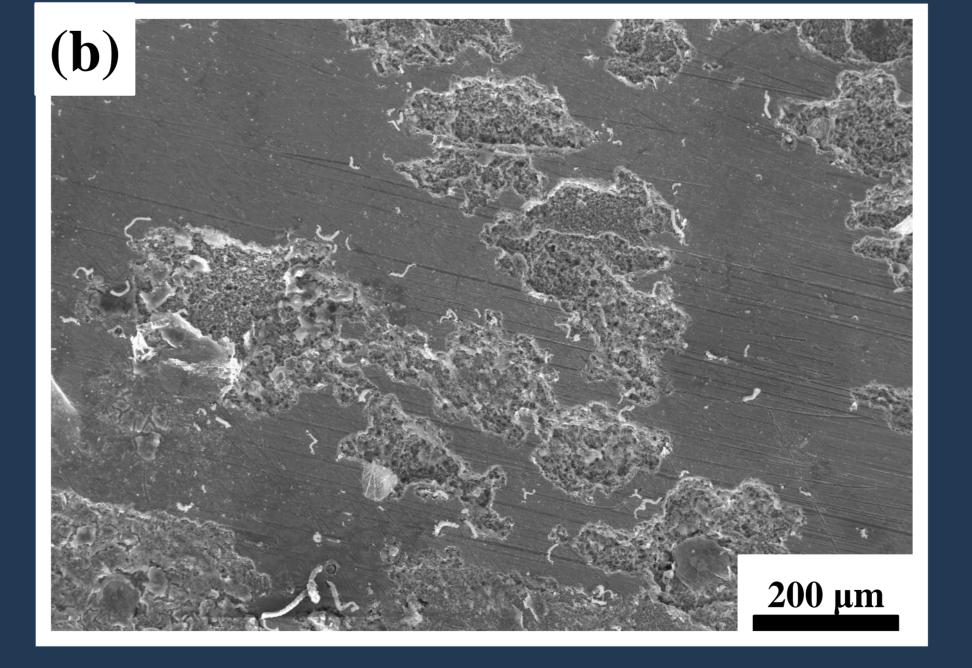




**Figure 4** Potential polarization curves of as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  metallic glasses and its partial crystalline alloys at 310 K in Hank's solution.

**Table I** Corrosion properties of as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  metallic glasses and its partial crystalline alloys at 310 K in Hank's solution





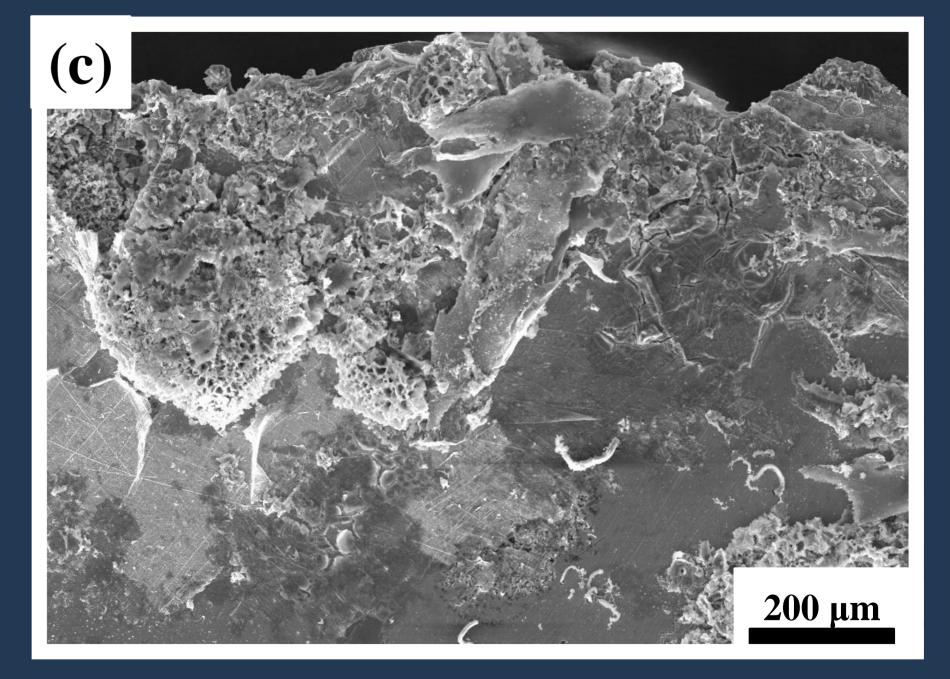


Figure 5 SEM images of (a) as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  metallic glasses and  $Zr_{53}Cu_{30}Ni_9Al_8$  metallic glasses annealed at (b) 712 K and (c) 722 K.

## Conclusions

- 1. The as-fabricated  $Zr_{53}Cu_{30}Ni_9Al_8$  exhibits lower corrosion current density ( $I_{corr}$ ), higher corrosion current potential ( $E_{corr}$ ), lower spontaneous passive current density, and higher pitting overpotential ( $E_{pit}$ - $E_{corr}$ ) than its partial crystalline alloys.
- 2. According to the results as mentioned above, the as-fabricated Zr<sub>53</sub>Cu<sub>30</sub>Ni<sub>9</sub>Al<sub>8</sub> possesses better corrosion resistance, and it is much easier to form the protective passive layer for avoiding the pitting reaction by chloride in SBF.